COURSE CURRICULUM

for

B.TECH. DEGREE

in

COMPUTER SCIENCE & ENGINEERING (DATA SCIENCE)

(Applicable from the academic session 2024-2025)



Dr. B. C. Roy Engineering College

An Autonomous Institution

Approved by: All India Council for Technical Education (AICTE)

Affiliated to: Maulana Abul Kalam Azad University of Technology, West Bengal (Formerly Known as -WBUT)

Jemua Road, Durgapur, West Bengal, India, 713206

The first year syllabus is unanimously accepted and approved in the first BoS meeting held in the Department of a) Physics, b) Chemistry, c) Mathematics, d) English, e) Electrical Engineering, f) Electronics and Communication Engineering, g) Computer Science and Engineering, h) Mechanical Engineering.



Head of the Department
Computer Science & Engineering (DS)
Dr. B. C. Roy Engineering College
Durgapur

A. Chandan Bandyopodhyay

Signature of the BoS Chairman, CSE (DS)



Course Name: Mathematics-I Course Code: BSC-M 101 (Semester- I)

Course Broad Category: Basic Science

1. Course Prerequisite:

Concept of Mathematics in 10+2 standard.

2. Course Learning Objectives:

The objective of this courses to familiarize the prospective engineers with techniques in calculus multivariate analysis and linear algebra. It aims to equip the students with standard concepts and tools at an intermediate to advance level that will serve them well towards tackling more advance level of mathematics and applications that they would find useful in their disciplines.

3. Teaching methodology and evaluation system for the course:

Teaching methodology – Lectures and Presentations, Interactive Discussions and Case Studies, Guest Lectures.

Evaluation System -

- A. Mid-Term Exam (20 Marks)- Summative Assessment (CIA-1)
- B. Internal Assessment (20 Marks)- Formative Continuous Assessment [Continuous Assessment 1 (CIA-2)]
- C. End-Semester Exam (60 Marks)- Summative Assessment.

4. Course Content:

Course Name: Mathematics-I Course Code: BSC-M 101 Hours per Week: 3L: 0T: 0P

Module	Topics	45L			
1.	Sequence and Series:	8L			
	Sequences: Limit of a Sequence, Boundedness, Convergence,				
	Divergence and Oscillation of a Sequence. Infinite Series: Necessary				
	Condition for Convergence, Tests for Convergence and Divergence,				
	Comparison Test: Only for Series with Positive Terms, Cauchy's				
	Integral Test, D Alembert's Ratio Test, Cauchy's Root Test,				

	Raabe's Test (Higher Ratio Test), Logarithmic Test, Alternating Series Leibnitz's Theorem, Absolute Convergence and Conditional Convergence, Power Series.	
2.	Differential Calculus (Single Variables): Functions of single variable: Rolle's Theorem, Lagrange's Mean Value Theorem, Cauchy's Mean Value Theorem, Taylor's Series and Maclaurin's Series Expansions with Different forms of Remainders (Only statements and applications in all cases).	6L
3.	Differential Calculus (Several Variables): Functions of Several Variables: Limit and Continuity, Partial Differentiation, Total Derivative. Partial Differentiation of Composite Functions: Change of Variables, Differentiation of an Implicit Function, Euler's Theorem for Two variables (statement only), Jacobian, Taylor's Theorem for Function of Two Variables (statement only), Maxima and Minima of Functions of Two Variables: With and Without Constraints, Lagrange's Method of Undetermined Multipliers.	10L
4.	Integral Calculus: Improper Integrals: Different types of Improper Integrals and their Convergence, Beta and Gamma function and their properties. Double Integrals and Triple Integrals - Application of Double Integrals and Triple Integrals, Change of Order of Integration.	9L
5.	Linear Algebra: Real Matrices: Symmetric, Skew-Symmetric, and Orthogonal Matrices (examples and uses), Inverse of a Matrix, Rank of a Matrix, Diagonalization of a Matrix. Complex Matrices: Hermitian, Skew-Hermitian, Unitary Matrices (examples and uses). Determinants, Solutions of Linear System of Equations Existence, Uniqueness. Cramer's Rule, Gauss-Jordan Elimination., Gauss Elimination Method, LU-decomposition Method from Gaussian Elimination. Basics of Group Theory. Vector Spaces, Basis, Dimension and Nullity. Linear Dependence and Independence. Matrix Eigenvalue Problem - Determining Eigenvalues and Eigenvectors, Cayley-Hamilton Theorem (statement only and uses), Some Applications of Eigenvalue Problems, Linear Transformations.	12L

Text Book:

- B. S. Grewal-- Higher Engineering Mathematics; **Publisher.** Khanna **Publishers.**
- Ramana B. V. --- Higher Engineering Mathematics; **Publisher.** McGraw Hill Education.

Reference Books:

- G. B. Thomas & R.L. Finney--- Calculus and Analytic Geometry; **Publisher.** Penguin Random House Australia.
- E. Kreyszig--- Advanced Engineering Mathematics (9th Edition); **Publisher.** Wiley.
- Veerarajan T.--- Engineering Mathematics for 1st year (TMG); **Publisher.** McGraw Hill Education.
- D. Poole---- Linear Algebra: A Modern Introduction; **Publisher.** Cengage Learning India Private Limited.
- Bali & Goel--- Text Book of Engineering Mathematics; **Publisher**. Laxmi **Publications**
- H. K. Das---Higher Engineering Mathematics; **Publisher**. Visionias.

6. Course Outcomes (CO):

Course	Details/Statement	Action	Knowledge
Outcomes		Verb	Level
BSC-M 101.1	Remember to identify different tools in	Identify	Remember
	algebra and calculus which would enable		
	them to devise engineering solutions to		
	encounter in their profession life.		
BSC-M 101.2	Understand the concept to explain	Explain	Understand
	applications of functions of single and		
	several variables in applied sciences and		
	engineering problems.		
BSC-M 101.3	Apply to implement the concept of partial	Implement	Apply
	derivatives in finding the maxima and		
	minima of a function of several variables		
	in the area of real-life problems.		
BSC-M 101.4	Analyze the ideas of mentioned	Organize	Analyze
	mathematical tools to organize complex		
	real-life problems.		
BSC-M 101.5	Evaluating the gradation of described	Assess	Evaluate
	mathematical tools in linear algebra to		
	assess the right approach to solve		
	multidisciplinary engineering problems.		
BSC-M 101.6	Construct logical and analytical skills to	Construct	Create
	create a new idea appreciated by		
	academics, research & emerging trends in		
	industry.		

7. Mapping of course outcomes to module / course content

Module	CO1	CO2	CO3	CO4	CO5	CO6
1	3	-	-	2	-	1
2	2	3	-	1	-	1
3	2	3	3	2	-	1
4	3	-	-	2	-	1
5	2	-	-	1	3	1

8. Mapping of the Course outcomes to Program Outcomes (PO)

	PO1	PO	PO1	PO1	PO1							
		2	3	4	5	6	7	8	9	0	1	2
CO1	2	2	1	1	2	-	-	-	-	-	-	1
CO2	1	2	3	1	-	-	-	-	-	-	-	1
CO3	1	2	2	1	1	-	-	-	-	-	-	1
CO4	1	2	1	1	2	-	-	-	-	-	-	2
CO5	2	2	2	2	3	-	-	-	-	-	-	1
CO6	1	1	1	1	-	-	-	-	-	-	-	1

	PSO1	PSO2	PSO3	PSO4
CO1	2	1	2	3
CO2	2	3	3	3
CO3	2	2	2	3
CO4	1	2	3	3
CO5	2	2	2	3
CO6	3	3	3	2



Course Name: CHEMISTRY
Course Code: BSC-CH 101
(Semester- I)

Course Broad Category: BASIC SCIENCE

1. Course Prerequisite:

Class-XII level knowledge of Chemistry, Mathematics, and Environmental Science.

2. Course Learning Objectives:

- i. This course impart basic knowledge of chemistry, general awareness of environmental pollution aspects, their impact and the basic ideas about material science (structure property relationships) among the engineering students for the better foundation of technical education.
- ii. To provide basic fundamentals among the upcoming young engineers and to carry out advanced research work in chemistry and allied domains.

3. Teaching methodology and evaluation system for the course:

Teaching methodology –Lectures and Presentations, Interactive Discussions and Case Studies and Guest Lectures.

Evaluation System –

- A. Mid-Term Exam (20 Marks)- Summative Assessment (CIA-1)
- B. Internal Assessment (20 Marks)- Formative Continuous Assessment [Continuous Assessment 1 (CIA-2)]
- C. End-Semester Exam (60 Marks)- Summative Assessment.

4. Course Content:

Course Name: Chemistry Course Code: BSC-CH 101 Hours per Week: 3L: 0T: 0P

Module	Topics	42L
1.	<u>Chemical Bonding:</u> Molecular orbitals of homonuclear and heteronuclear	8L
	diatomic molecules. Equations for atomic and molecular orbitals. Energy	

Topics	42L
level diagrams of diatomics. Crystal field theory and the energy level	
diagrams for transition metal ions and their magnetic properties.	
Organometallic compounds and their Industrial Applications. Band	
structure of solid sand the role of doping on band structure.	
Thermodynamics & Electrochemistry: Introduction, thermodynamic	7L
functions: enthalpy, entropy and free energy. Estimations of entropy and	
free energies. Gibbs Helmholtz equation.	
Electrochemistry- conductance, Sp. Conductance, equivalent conductance,	
Ostwald dilution law, conductometric titration. Free energy and EMF.	
Cellpotentials, The Nernst equation and applications. Latium Ion	
Battery,H ₂ /O ₂ fuel Cells. Water chemistry. Corrosion.	
Organic Reactions & Mechanism: Introduction to reactions involving	6L
substitution,, addition, elimination, oxidation, reduction, Condensation.	
	5L
industrial applications. Polymer molecular weight (number avg. weight	
avg. viscosity avg.: Theory and mathematical expression only), Poly	
	6L
	OL
Tron and steet managacturing, Hardening and annealing, Alloy Steet	
Spectroscopic Techniques & Applications: Basic concepts of HV IR	3L
	JL
	3L
	ЭL
environmental application like water purification.	
Composite materials: Introduction, Constituents, Fibrerein forced	
	level diagrams of diatomics. Crystal field theory and the energy level diagrams for transition metal ions and their magnetic properties. Organometallic compounds and their Industrial Applications. Band structure of solid sand the role of doping on band structure. Thermodynamics & Electrochemistry: Introduction, thermodynamic functions: enthalpy, entropy and free energy. Estimations of entropy and free energies. Gibbs Helmholtz equation. Electrochemistry- conductance, Sp. Conductance, equivalent conductance, Ostwald dilution law, conductometric titration. Free energy and EMF. Cellpotentials, The Nernst equation and applications. Latium Ion Battery, Hz/Oz fuel Cells. Water chemistry. Corrosion. Organic Reactions & Mechanism: Introduction to reactions involving substitution,, addition, elimination, oxidation, reduction, Condensation. Polymer: Properties & Applications: Concepts, classifications and industrial applications. Polymer molecular weight (number avg. weight avg. viscosity avg.: Theory and mathematical expression only), Poly dispersity index (PDI). Polymerization processes (addition and condensation polymerization), degree of polymerization, Copolymerization, stereo-regularity ofpolymer, crystallinity (concept of Tm) and amorphicity (Concept of Tg) of polymer. Preparation, structure and use of some common polymers: plastic (PE:HDPE, LDPE, PVC, Bakelite, PP), rubber (natural rubber, SBR, NBR) and Vulcanization. Fibre (nylon 6.6, Nylon 6, Polyester). Conducting and semi-conducting polymers. Industrial Chemistry: Solid Fuel: Coal, Classification of coal, constituents of coal, carbonization of coal (HTC and LTC), Coalanalysis: Proximate and ultimate analysis. Liquid fuel: Petroleum, classification of petroleum, Refining, Petroleum distillation, Thermal cracking, Octane number, Cetane number, Gaseousfuels: Naturalgas, watergas, coalgas, biogas. Cement and its classification, Chemical composition, Physical properties, Setting and hardening, Properties. Iron and steel manufacturing, Hardening and anneal

Module	Topics	42L
	Green Chemistry:	3L
	Introduction, Principles of green chemistry, Use of alternative feed stock,	
	Use of Alternative Solvents, Design of safer chemicals, Bio-fuel for	
	sustainable development using green chemistry. Atom economy.	

Text Book:

- 1. Engineering Chemistry, JainandJain ,Dhanpat Rai & Co Pvt Ltd
- 2. A Text book of Engineering Chemistry, Shashi Chawla, Dhanpat Rai & Co PvtLtd
- 3. Engineering Chemistry, Gourkrishna Dasmohapatra, Vikas Publishing House
- 4. Engineering Chemistry, K. L. Chugh, Kalyani Publishers.

Engineering Chemistry, Willey

Reference Books:

- 1. General & Inorganic Chemistry, R. P. Sarkar, New Central Book Agency P Ltd
- 2. I.L. Finar, Organic Chemistry, Addison Wesley Longman, Inc
- 3. Organic Chemistry, Morrison & Boyd, Prentice Hall of India
- 4. Physical Chemistry, K.L. Kapoor, McMillan
- 5. P. C. Rakshit, Physical Chemistry, Sarat Book House (7thEdition)
- 6. Green Chemistry Theory and Practice By Paul T. Anastas, John Charles Warner Oxford University Press

: 9788126543205

6. Course Outcomes (CO):

Course	Details/Statement	Action	Knowledge
Outcomes		Verb	Level
BSCH101.1	Correlate Structure and Properties of Solids	Identify,	Analyse
	and Coordination Compounds.	Select	
BSCH101.2	Rationalisebulkpropertiesandprocessesusingt	Explain	Understand
	hermodynamicandelectrochemicalconcept		
BSCH101.3	Interaction of radiation with matter and	Recognize	Understand
	structural elucidation.		
BSCH101.4	Analysis of polymeric property and	Identify,	Apply
	application of industrial materials.	Implement	
BSCH101.5	Importance of green chemistry and	Design	Apply
	nanomaterial towards structural application		
	and sustainable development.		
BSCH101.6	Organic reaction mechanism and structural	Implement	Apply

determination.	

7. Mapping of course outcomes to module / course content

Module	CO1	CO2	CO3	CO4	CO5	CO6
1	3	3	-	-	-	-
2	-	-	-	3	-	3
3	-	-	3	2	-	-
4	-	-	-	-	3	-

8. Mapping of the Course outcomes to Program Outcomes (PO)

	PO1	PO	PO1	PO1	PO1							
		2	3	4	5	6	7	8	9	0	1	2
CO1	3	2	-	1	-	-	1	-	2	1	-	3
CO2	3	3	-	1	-	-	1	-	2	1	-	3
CO3	3	2	-	2	-	-	1	-	2	1	-	3
CO4	3	2	-	1	-	-	1	-	2	1	-	3
CO5	3	2	-	1	-	-	2	-	2	1	-	3
CO6	3	2	-	1	-	-	1	-	2	1	-	3

9. Mapping to Program Specific Outcomes (PSO)

	PSO1	PSO2	PSO3	PSO4
CO1	2	2	1	2
CO2	2	1	1	2
CO3	2	2	2	1
CO4	1	1	1	2
CO5	2	2	1	2
CO6	1	1	1	1

*** End of Syllabus***



Course Name: BASIC ELECTRONICS ENGINEERING Course Code: ESC-EC 101 (Semester– I)

Course Broad Category: ENGINEERING SCIENCE

1. Course Prerequisite:

Class-XII level knowledge of Physics and Mathematics

2. Course Learning Objectives:

- i. This course introduces the concepts about solid-state electronic components and their applications.
- ii. Students will also learn to design and analyze basic analog electronic and digital logic circuits.

3. Teaching methodology and evaluation system for the course:

Teaching methodology – Lectures and Presentations, Interactive Discussions and Case Studies.

Evaluation System -

- A. Mid-Term Exam (20 Marks)- Summative Assessment (CIA-1)
- B. Internal Assessment (20 Marks)- Formative Continuous Assessment [Continuous Assessment (CIA-2)]
- C. End-Semester Exam (60 Marks)- Summative Assessment.

4. Course Content:

Course Name: BASIC ELECTRONICS ENGINEERING

Course Code: ESC-EC 101 Hours per Week: 3L: 0T: 0P

Module	Topics	36L
1.	Semiconductor Diodes: Semiconductor materials- intrinsic and extrinsic types • Ideal Diode, practical diode: open circuit, forward and reverse bias condition, V-I characteristics, knee voltage, junction breakdown: Avalanche and Zener • Peak Inverse Voltage (PIV), small signal model, Zener diode and its application • Halfwave, Full-wave (centre tapped) and bridge Rectifier: efficiency, ripple factor, PIV etc, Clipper: positive, negative and biased, Clamper: positive and negative, biased	11L
2.	Bipolar Junction Transistors (BJTs): Physical structure and operation• Operating region of BJT • D.C. analysis • Biasing the BJT: fixed bias, emitter feedback bias, collector feedback bias and voltage divider bias• load line, Bias stability, Basic BJT amplifier configuration: common emitter, common base and common collector • relation between α , β and γ , Transistor as a switch: cut-off and saturation modes.	10L
3.	MOS Field Effect Transistor (MOSFET): Enhancement-type MOSFET: structure and physical operation, •Depletion, accumulation and inversion region, •threshold potential, Pass-characteristics •Drain and Transfer characteristics, •Cut-off, linear and saturation region, punch-through breakdown • drain resistance, trans-conductance, •channel length modulation, •Depletion-type MOSFET• CMOS Inverter, transient response, delay definition.	5L
4.	Operation Amplifier (Op-amps): Ideal Op-amp Characteristics: Common Mode Rejection Ratio (CMRR), input/output offset voltage and current, slew rate, PSRR etc• Op-amp circuits: comparator, unity gain buffer, inverting amplifier, non-inverting amplifier, adder, subtractor, integrator, differentiator etc.	5L
5.	Digital Electronics: Binary, octal and hexadecimal number systems, number conversion. Binary arithmetic: Addition, subtraction, multiplication and division, basic and universal Logic gates, Truth tables. Function Representation in SOP, POS form; Min-term, Max-term, Canonical Form, Boolean algebra, demorgan's theorem, k-map optimization, logic design.	5L

Text Book:

- Electronic Principles, by Sanjay Sharma, S.K.Kataria & Sons
- Integrated Electronics by Millman & Halkias

Reference Books:

• Electronics Devices & Circuits by Salivahanan

• Electronics Circuits by Schilling & Belove

6. Course Outcomes (CO):

After going through this course the Students will be able to:

Course	Details/Statement	Action	Knowledge
Outcomes		Verb	Level
ESC-	Understand and Explain the characteristics	Explain	Understand
EC101.1	of P-N junction diode		
ESC-	Identify and analyze the characteristics of	Identify	Analyze
EC101.2	Bipolar Junction Transistor		
ESC-	Identify the characteristics of MOSFET and	Identify,	Apply
EC101.3	apply for digital circuit design	Design	
ESC-	Apply diode, BJT and MOSFET to design	Design	Apply
EC101.4	various electronic circuits		
ESC-	Understand and Implement op-amp circuit	Implement	Understand
EC101.5	for different mathematical operations		
ESC-	Design and analyze digital logic circuits	Design	Analyze
EC101.6			

7. Mapping of Course Outcomes (CO) to module / course content

Module	CO1	CO2	CO3	CO4	CO5	CO6
1	3	-	-	2	-	-
2	-	3	-	2	-	-
3	-	-	3	2	-	-
4	-	-	-	2	3	-
5	-	-	-	2	-	3

8. Mapping of the Course Outcomes (CO) to Program Outcomes (PO)

	PO1	PO	PO1	PO1	PO1							
		2	3	4	5	6	7	8	9	0	1	2
CO1	3	1	1	1	-	-	-	-	-	2	-	1
CO2	3	1	1	1	-	-	-	-	-	2	-	1
CO3	3	1	1	1	-	-	-	-	-	2	-	1
CO4	3	1	1	1	-	-	-	-	-	2	-	1
CO5	3	1	1	1	-	-	-	-	-	2	-	1

	PSO1	PSO2	PSO3	PSO4
CO1	1	2	1	1
CO2	2	1	1	1
CO3	1	1	1	1
CO4	1	2	1	1
CO5	1	1	1	1
CO6	1	2	1	1



Course Name: ENGINEERING MECHANICS Course Code: ESC-ME 101 (Semester- I)

Course Broad Category: Engineering Science

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1. Course Prerequisite:

Class XI and XII level knowledge of Mechanics, Mathematics (Vector Algebra and Coordinate Geometry, Differential and Integral Calculus)

2. Course Learning Objectives:

- iii. This course introduces Statics and Dynamics of Engineering Mechanics with elaborate concepts on equilibrium of forces and motion under forces.
- iv. Students will also get introduced to the subject of Mechanics of Deformable Bodies.

3. Teaching methodology and evaluation system for the course:

Teaching methodology –Lectures and Presentations, Interactive Discussions and real world problem discussion.

Evaluation System –

- A. Mid-Term Exam (20 Marks)- Summative Assessment (CIA-1)
- B. Internal Assessment (20 Marks)- Formative Continuous Assessment [Continuous Assessment 1 (CIA-2)]
- C. End-Semester Exam (60 Marks)- Summative Assessment.

4. Course Content:

Course Name: Engineering Mechanics

Course Code: ESC-ME101 Hours per Week: 3L: 0T: 0P

Module	Topics	42 L	CO					
1.	Introduction to Statics: Importance of Mechanics in	12L	CO1					
	engineering; Fundamental idealization: Particle and Rigid body		and					
	concept; Types of forces (collinear, concurrent, parallel,		CO2					
	concentrated and distributed); Vector and scalar quantities;							
	Transmissibility of a force (sliding vector); Lami's Theorem.							
	Introduction to Vector Algebra: Parallelogram law; Addition							
	and subtraction of vectors; Free vector, Bound Vector;							
	Representation of Forces and Moments in terms of i, j, k; Cross							
	product and Dot product and their applications. Two and Three	product and Dot product and their applications. Two and Three						
	Dimensional Force Systems: Moment and Couple; Varignon's							
	theorem; Resolution of a coplanar force by its equivalent							

Module	Topics	42L	CO
	Force-couple system. Concept of Equilibrium in Two and Three dimensions; Free body concept; Equations of Equilibrium.		
2.	Concept of Friction: Laws of Coulomb friction; Angle of Repose; Coefficient of friction; Types of friction. Distributed Force: Centroid and Centre of Gravity; Centroids of a triangle, circular arc, quadrilaterals, circular sector, triangular lamina, composite areas. Moment of Inertia: M.I of plane figure with respect to an axis in its plane; M.I of plane figure with respect to an axis perpendicular to the plane of the figure, Parallel axis theorem; Mass moment of inertia of symmetrical bodies, e.g. circular plate, ring, cylinder, sphere, rod.	12L	CO3 and CO4
3.	Introduction to Dynamics: Kinematics and Kinetics; Newton's laws of motion; Law of gravitation & acceleration due to gravity; Rectilinear motion of particles; determination of position, velocity and acceleration under uniform and non uniformly accelerated rectilinear motion; construction of x-t, v-t and a-t graphs. Plane Curvilinear Motion of Particles: Rectangular components (Projectile motion); Normal and Tangential components (Circular motion). Kinetics of Particles and Rigid bodies: Newton's second law; Equation of motion; D. Alembert's principle and free body diagram; Principle of work and energy; Principle of conservation of energy; Principle of Linear Impulse and Momentum; Power and efficiency with simple examples.	16L	CO5
4.	Concept of Simple Stresses and Strains: Normal stress, Shear stress, Bearing stress, Normal strain, Shear strain; Hooke's law; Poisson's ratio; Stress-strain diagram of ductile and brittle materials; Proportional limit; Elastic limit; Yielding; Ultimate stress; Modulus of elasticity; Working stress; Factor of safety; Definition of malleability, ductility, toughness and resilience; Basic concept of thermal stress and strain.	5L	CO6

Text Book:

1. Engineering Mechanics by S. S. Bhavikatti, New Age International Publishers, 9th Edition, 2023.

2. Engineering Mechanics by S. Rajasekaran & G. Sankarasubramanian, Vikas Publishing House Pvt Ltd., Noida.

Reference Books:

- 1. Engineering Mechanics: Statics & Dynamics by I. H. Shames, 4th ed. PHI.
- 2. Engineering Mechanics: Statics & Dynamics by Hibbeler & Gupta, 11th ed. Pearson.
- 3. Engineering Mechanics by R.S. Khurmi, S. Chand Publications, Delhi.
- 4. A Text Book of Engineering Mechanics by Bansal R. K.(2010), Laxmi Publications.
- 5. Elements of Strength of Materials by Timoshenko & Young, 5th ed. E.W.P.
- 6. Engineering Mechanics by M.P. Poonia & D.S. Bedi, Khanna Publishing House, 2019.
- 7. Mechanics for Engineering by Beer, F.P. and Johnston.
- 8. Fundamental Concepts In Engineering Mechanics by Suman Chakraborty, Everest PublishingHouse
- 9. Engineering Mechanics by Timoshenko, Young and Rao, Revised 4th ed. TMH.
- 10. Engineering Mechanics by Basudeb Bhattacharyya- Oxford University Press.
- 11. Fundamentals of Engineering Mechanics by Debabrata Nag & Abhijit Chanda ChhayaPrakashani
- 12. Engineering Mechanics [Vol-I & II] by Meriam & Kraige, 5th ed. Wiley India.

6. Course Outcomes (CO):

Course	Details/Statement	Action	Knowledge
Outcomes		Verb	Level
CO1	Explain the fundamental principles of mechanics, including the importance of mechanics in engineering and the concepts of particles and rigid bodies.	Explain	Understand
CO2	Determine analytically the forces and moments acting on statically determinate structure.	Identify, Select	Understand
CO3	Understand the presence and effects of friction in statics and dynamics.	Implement	Understand
CO4	Compute the centroid of lamina and centre of gravity of solids, area moment of inertia, mass moment inertia of solids.	Compute	Evaluate
CO5	Apply the fundamental concept of kinematics and kinetics to determine displacement, velocity and acceleration of particle and rigid body during linear/ angular/ general plane motions and the extension of Newton's law of motion to work	Apply	Apply

	energy principle and Impulse momentum principle.		
CO6	Understand the deformation and mechanical behaviour of bodies under various load conditions.	Recognize	Understand

7. Mapping of course outcomes to module / course content (hrs)

Module	CO1	CO2	CO3	CO4	CO5	CO6
1	6	6				
2			6	6		
3					16	
4						5

8. Mapping of the Course outcomes to Program Outcomes (PO)

	PO1	PO	PO1	PO1	PO1							
		2	3	4	5	6	7	8	9	0	1	2
CO1	2	1	1	1	1					1		1
CO2	3	2	1	1	1					1		1
CO3	3	1	1	1			1					1
CO4	2	2	1	1	1	1				1		1
CO5	2	1	1	1	1	1			1	1		1
CO6	3	2	1	1	2	2						1

	PSO1	PSO	PSO	PSO
		2	3	4
CO1	2	2	3	2
CO2	3	3	1	1
CO3	1	2	1	1
CO4	1	2	1	1
CO5	1	2	1	1
CO6	1	1	1	1



Course Name: Introduction to Computer Hardware and Software Course Code: ESC-CS 101 (Semester– I)

Course Broad Category: Engineering Science

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1. Course Prerequisite:

Basic knowledge of computers and general mathematical operations.

2. Course Learning Objectives:

- i. To introduce students to the fundamental concepts of computer hardware and software.
- ii. To equip students with knowledge and practical skills essential for understanding modern computing systems and networks.
- iii. To provide an overview of emerging computing technologies and their applications in various domains.

3. Teaching methodology and evaluation system for the course:

Teaching methodology –Lectures and Presentations, Interactive Discussions and Case Studies.

Evaluation System -

- A. Mid-Term Exam (20 Marks)- Summative Assessment (CIA-1)
- B. Internal Assessment (20 Marks)- Formative Continuous Assessment [Continuous Assessment 1 (CIA-2)]
- C. End-Semester Exam (60 Marks)- Summative Assessment.

4. Course Content:

Course Name: Introduction to Computer Hardware and Software Course

Course Code: ESC-CS 101 Hours per Week: 3L: 0T: 0P

Module	Topics	Lectures
Unit 1:	Definition and Characteristics of Computers, Evolution of	4L
Computer	Computers, Generations of Computers, Classification of	
Basics	Computers, Applications of Computers, Computer System, its	

	Components, and their Functions.	
Unit 2: Number	Basic Concepts of Number Systems, Binary, Octal, Decimal,	10L
Systems and	and Hexadecimal Number Systems, Conversion Between	
Data	Different Number Systems (Base 2 to Base 10), Shortcut	
Representation	Methods for Conversion among Binary, Octal, and	
	Hexadecimal, Arithmetic Operations in Different Number	
	Systems, Signed and Unsigned Number Systems, Data	
	Representation: Bits, Bytes, and Words, ASCII and Unicode	
	Character Sets, Boolean Algebra and Logic Gates	
Unit 3: Memory	Introduction to Computer Memory, Understanding Data	4L
	Representation in Computer Memory, Speed, Capacity, and	
	Memory Hierarchy, Primary Memory and its Types,	
	Secondary Memory and Classification of Secondary Memory,	
	Various Secondary Storage Devices and Their Utilities.	
Unit 4:	Classification of Computer Software, System Software and	4L
Computer	Application Software, Utility Software, Operating System	
Software and its	(OS): Role and Types (CLI-based OS, GUI-based OS),	
Types	Mobile OS and Network OS.	
Unit 5: Data	Fundamental Concepts of Data Communication, Transmission	6L
Communication,	Media, Multiplexing and Switching, Computer Networks	
Computer	(LAN, WAN, MAN, PAN), Network Topologies (Bus, Star,	
Network, and	Ring, Mesh, and Hybrid), Network Devices: Router, Switch,	
Internet Basics	Hub, Bridge, Gateway, Basics of the Internet and Protocols	
	(TCP/IP, HTTP, FTP, etc.), Email and Client-Server	
	Architecture.	
Unit 6: Basics	Problem-Solving: From Algorithms (Pseudo code and	6L
of Programming	Flowcharts) to Program, Concept of Variables and Constants,	
	Operators, Naming Rules for Variables. Decision Making and	
	Iteration, Concept of Compilers and Interpreters.	
Unit 7:	Multi-core processor, Graphics Processing Unit (GPU),	4L
Emerging	Tensor Processing Unit (TPU), Neural Processing Unit (NPU),	
Processing and	Quantum Processor, Artificial Intelligence and Machine	
Computing	Learning (AI/ML), Cloud Computing, Cybersecurity, Internet	
Technologies	of Things (IoT), Augmented Reality (AR) and Virtual Reality	
	(VR), Big Data and Data Analytics.	

Text & References Books:

- Introduction to Computer Science; Pearson publication
- Fundamentals Of Computers by V. Rajaraman, PHI Learning
- Fundamentals of Computers by E Balagurusamy, McGraw Hill Education (India)

6. Course Outcomes (CO):

After going through this course the Students will be able to:

Course	Details/Statement	Action	Knowledge
Outcomes		Verb	Level
ESC-	Define, describe, and classify computers,	Define,	Remember,
CS101.1	their evolution, and their components.	describe,	Understand
	Explain the role of computer systems and	classify	
	their applications in real-world scenarios.	&Explain	
ESC-	Apply arithmetic operations and conversion	Apply,	
CS101.2	techniques between binary, octal, decimal,	Analyze,	
	and hexadecimal number systems. Analyze	Evaluate &	Apply,
	Boolean algebra expressions and evaluate	Solve	Analyze
	logic gate operations to solve problems in		
	data representation.		
ESC-	Analyze different memory types (primary	Analyze,	
CS101.3	and secondary) and their characteristics such	Compare	Analyze,
	as speed, capacity, and usage. Compare and	& Evaluate	Evaluate
	evaluate the performance impact of memory		Dvaraace
	hierarchy on computer systems.		
ESC-	Classify various types of software (system	Classify,	
CS101.4	software, application software, utility	Describe,	
	software) and describe their roles in the	Examine	Understand,
	functioning of computer systems. Examine	&Assess	Classify,
	operating systems, mobile OS, and network		Examine
	OS and assess their impact on the		
	environment and user experience.		
ESC-	Apply principles of data communication and	Apply,	Apply,
CS101.5	network design, including the use of routers,	Analyze, &	Analyze,
	switches, and other networking devices.	Demonstrat	Demonstrate
	Analyze network topologies, internet	e	
	protocols (TCP/IP, HTTP, FTP), and		
	demonstrate the use of networking tools to		
	establish basic computer networks.		
ESC-	Explore emerging technologies such as	Explore,	Evaluate,
CS101.6	multi-core processors, GPUs, and quantum	Evaluate,	Explore,
	processors. Evaluate the applications of	Demonstrat	Collaborate
	Artificial Intelligence, Cloud Computing,	e, Adapt &	
	IoT, AR/VR, and Big Data in modern	Collaborate	
	computing. Students will demonstrate the		
	ability to adapt to new technologies and		
	collaborate in interdisciplinary teams.		

7. Mapping of course outcomes to module / course content

Module	CO1	CO2	CO3	CO4	CO5	CO6
1	3	-	-	-	-	-
2	ı	3	1	1	1	-
3	-	-	3	-	-	-
4	-	-	-	3	-	-
5	-	-	-	-	3	-
6	-	-	-	-	-	3
7	-	-	-	-	-	3

8. Mapping of the Course outcomes to Program Outcomes (PO)

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
CO1	3	2	1	-	-	1	-	1	-	-	-	-
CO2	-	3	-	2	-	-	-	-	-	-	-	-
CO3	3	-	2	-	-	1	-	ı	-	-	-	-
CO4	-	-	-	-	-	2	2	-	-	-	-	-
CO5	-	3	-	-	3	-	-	1	-	-	-	-
CO6	-	-	-	-	-	-	-	-	3	-	-	3
AVG.	1.0	1.33	0.33	0.33	0.5	0.33	0.33	0.0	0.5	0.0	0.0	0.5

	PSO1	PSO2	PSO3	PSO4
CO1	3	3	3	3
CO2	2	2	2	3
CO3	3	2	3	3
CO4	3	2	2	2
CO5	3	3	3	3
CO6	3	3	3	3



Course Name: Environmental Science Course Code: MC-ES 201 (Semester- II)

Course Broad Category: Mandatory Course

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1. Course Prerequisite:

Class-XII level knowledge of Environmental Science, Chemistry and Mathematics.

2. Course Learning Objectives:

- i. This course impart basic knowledge of environment, general awareness of environmental pollution aspects, their impact and the basic ideas about material science (structure property relationships) among the engineering students for the better foundation of technical education.
- ii. To provide basic fundamentals among the upcoming young engineers and to carry out advanced research work in environmental science and allied domains.

3. Teaching methodology and evaluation system for the course:

Teaching methodology –Lectures and Presentations, Interactive Discussions and Case Studies and Guest Lectures.

Evaluation System-

- A. Mid-Term Exam (20 Marks)- Summative Assessment (CIA-1)
- B. Internal Assessment (20 Marks)- Formative Continuous Assessment [Continuous Assessment 1 (CIA-2)]
- C. End-Semester Exam (60 Marks)- Summative Assessment.

4. Course Content:

Course Name: Environmental Science Course Code: MCES- 101 / MCES- 201

Hours per Week: 1L:0T:0P

Module	Topics	42L				
1.	Basic Concepts of Environment: Basic ideas of environment and its	8L				
	component, Renewable and nonrenewable resources and reserves.					
	Air pollution and control: Atmospheric Composition: Troposphere,					
	Stratosphere, Mesosphere, Thermosphere and their temperature					
	variations.					
	Lapse rate: Ambient lapse rate, adiabatic lapse rate, atmospheric					
	stability, temperature inversion (radiation inversion).					
	Green House Effects: Definition, impact of greenhouse gases Green					
	house effects: Definition, impact of green house gases on the global					
	climate and consequently on sea water level, agriculture and marine					
	food. Global warming and its consequence, Control of Global					
	warming. Acid rain: causes, effects and control. Earth's heat budget,					
	carbon capture, carbon footprint					
	Definition of pollutants and contaminants, Primary and secondary					
	pollutants, criteria pollutant. Sources and effect of different air					
	pollutants- Suspended particulate matter, oxides of carbon, oxides of					
	nitrogen, oxides of sulphur, particulate, PAN. Photochemical Smog.					
	Standards and control measures: Industrial, commercial and					
	residential air quality standard, control measure (ESP, cyclone					
	separator, bag house, catalytic converter, scrubber					
	Depletion Ozone layer: CFC, destruction of					
	ozone layer by CFC, impact of other green					
	house gases, effect of ozone modification					
2.	Water Pollution and Control	4L				
	Pollutants of water, their origin and effects:					
	Oxygen demanding wastes, pathogens, nutrients,					
	Salts, thermal application, heavy metals,					
	pesticides. 2L					
	River/Lake/groundwaterpollution:River:DO,5dayBODtest,					
	Unseeded and Seeded BOD test, BOD reaction rate constants,					
	COD.					
	Aquifers, hydraulic gradient, ground water flow (Definition only)					
6.	Noise Pollution	2L				
	Definition of noise, effect of noise pollution, noise classification					
	[Transportnoise, occupational noise, neighborhood noise]. Definition					
	of noise frequency, noise pressure, noise intensity, noise threshold					
	limit value, equivalent noise level, L ₁₀ (18hr Index), effective					
	perceived noise level. Noise pollution control.					
4.	Land Pollution	2L				
	Solid Waste: Municipal, industrial, commercial, agricultural, domestic,					
	pathological and hazardous solid wastes, electronic waste					
	Recovery and disposal method- Open dumping, Land filling,					

Module	Topics	42L
	incineration, composting. Recycling, Reduce, Reuse, Refuse.	

Text Book:

1. Gour Krishna Das Mahapatra, Basic Environmental Engineering and Elementary Biology, Vikas Publishing House P. Ltd.

Reference Books:

- 1. Masters, G. M., "Introduction to Environmental Engineering and
- 2. Science", Prentice-Hal lof India Pvt. Ltd., 1991.
 - 3. De, A. K., "Environmental Chemistry", New Age International.

6. Course Outcomes (CO):

Course	Details/Statement	Action	Knowledge
Outcomes		Verb	Level
MCES-	To understand the natural environment and	Identify,	Understand
101.1MCES-	its relationships with human activities.	Select	
201.1			
MCES-	To apply the fundamental knowledge of	Apply,	Evaluate
101.2MCES-	science and engineering to assess	Explain	
201.2	environmental and health risk.		
MCES-	To develop guidelines and procedures for	Recognize	Create
101.3MCES-	health and safety issues obeying the		
201.3	environmental energy aspects.		
MCES-	To acquire skills for scientific problem-	Identify,	Analyze
101.4MCES-	solving related to air, water, land and noise	Implement	
201.4	pollution.		
MCES-	To acquire knowledge about various Waste	Design	Analyze
101.5MCES-	Management aspects.		
201.5			

7. Mapping of course outcomes to module / course content

Module	CO1	CO2	CO3	CO4	CO5
1	3	3	2	3	2
2	-	1	2	3	1
3	-	1	-	3	1
4	-	1	-	3	1

8. Mapping of the Course outcomes to Program Outcomes (PO)

	PO1	PO	PO1	PO1	PO1							
		2	3	4	5	6	7	8	9	0	1	2
CO1	1	-	-	-	-	2	2	2	-	1	-	2
CO2	1	-	-	-	-	2	2	2	-	1	-	2
CO3	1	-	-	-	-	2	2	2	-	1	-	2
CO4	1	-	-	-	-	2	2	2	-	1	-	2
CO5	1	-	-	-	-	2	2	2	-	1	-	2

	PSO1	PSO2	PSO3	PSO4
CO1	1	2	1	2
CO2	2	1	2	2
CO3	2	2	2	2
CO4	2	1	1	2
CO5	2	1	2	1



Course Name: CHEMISTRY LAB
Course Code: BSC-CH 191
(Semester – I)

Course Broad Category: BASIC SCIENCE

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1. Course Prerequisite:

Class-XII level knowledge of Chemistry Practical.

2. Course Learning Objectives:

- i. Expose students to various experimental skills and tools
- ii. Togainpracticalknowledgebyapplyingexperimentalmethodstocorrelatewiththe theory. Apply the analytical techniques and graphical analysis to the experimental data.

3. Teaching methodology and evaluation system for the course:

Teaching methodology: Instruction: This method recognizes that students have different learning styles, abilities, and backgrounds, and aims to create a learning environment that accommodates these differences.

Evaluation System-

Internal Assessment (60Marks)-Formative Continuous Assessment [Continuous Assessment; Note Book (30 Marks), Viva Voce (20 Marks), Attendance (10 Marks)] End-Semester Exam (40 Marks)- Summative Assessment.

4. Course Content:

Course Name: CHEMISTRY LAB

Course Code: BSC-CH191 Hours per Week: 0L: 0T: 2P

Module	Topics	10P
1.	Instrumental Analysis:	2P
	1. Conductometric titration for determination of the strength of a given HCl solution by titration against a standard NaOH solution.	
	2. pH-metric titration for determination of strength of a given HCl solution against a standard NaOH solution	
Module	Topics	10P
2.	Estimation through Titrimetric Methods:	
	1. To determine chloride ion in a given water sample by Argentometric method.	2P
	2. Removal and estimation of hardness of water by Complex ometric titration.	
3.	Determination of Specific Property of a Solution:	3P
	 Determination of viscosity of Sugar Solution. Determination of surface tension of a sugar solution by drop count method. Determination of partition coefficient of a substance between 	
	two immiscible liquids.	
4.	Verification of Law & Equations:	
	Verification of Lambert Beer's Law using visible colorimeter Determination of Stefan's radiation constant.	3P
	 Study the adsorption of oxalic acid fromsolutiononactivatedcharcoalandexaminethevalid ityofFreundlich isotherm. 	
	Determination of dissolved oxygen present in a water sample	

Minimum of eight experiments to be performed taking at least one from each module mentioned above.

Textbooks

- Experiments in Applied ChemistryBy Dr. SunitaRatan.
- Quantitative Chemical Analysis By Arthur Vogel.

Reference Books

- Practical Chemistry by Dr. R. C. Bhattacharya.
- Practical Chemistry By S. Chand.

5. Course Outcomes (CO):

After going through this course the Students will be able to:

Course	Details/Statement	Action	Knowledge
Outcomes		Verb	Level
BSC-	ImplementInstrumentalanalyticalprocedurefo	Evaluate	Analyze
CH291.1	rtheenrichmentofmoderntechnical skill.		
BSC-CH	Rationalize inter molecular phenomena using	Identify,	Evaluate
191.2	thermodynamic considerations.	Select	
BSC-CH	Understandtitrimetricmethodsofwateranalysis	Implement	Apply
191.3	requiredforenvironmentalcontext		
BSC-CH	Concept of radiation matter interaction and	Design	Create
191.4	its application.		
BSC-CH	Evaluate different surface phenomena by	Identify,	Apply
191.5	adsorption technique.	Implement	
BSC-CH	Estimate essential parameter like oxygen in	Recognize	Understand
191.6	water by titrimetric method.		

6. Mapping of course outcomes to module/course content

Module	CO1	CO2	CO3	CO4	CO5	CO6
1	3	-	-	-	-	-
2	-	-	3	-	-	-
3	-	2	-	-	1	-
4	-	-	-	1	1	1

7. Mapping of the Course outcomes to Program Outcomes (PO)

	PO1	PO2	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
CO1	3	2		1	-	-	1	-	2	1	-	3
CO2	3	2		1	-	-	1	-	2	1	-	3
CO3	3	2		1	-	-	1	-	2	1	-	3

CO4	3	2	1	-	-	1	-	2	1	-	3
CO5	3	2	1	-	-	1	-	2	1	-	3
CO6	3	2	1	-	-	1	-	2	1	-	3

	PSO1	PSO2	PSO3	PSO4
CO1	2	2	1	2
CO2	2	1	1	2
CO3	2	2	2	1
CO4	1	1	1	2
CO5	2	2	1	2
CO6	1	1	1	1



Course Name: BASIC ELECTRONICS ENGINEERING LAB Course Code: ESC-EC 191 (Semester – I)

Course Broad Category: ENGINEERING SCIENCE

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1. Course Prerequisite:

Class-XII level knowledge of Physics Practical.

2. Course Learning Objectives:

- i. Expose students to experimental skills on electronic circuit design and analysis
- ii. To gain practical knowledge by applying experimental methods to correlate with the basic electronic theory.
- iii. To apply the analytical techniques and graphical analysis to the experimental data.

3. Teaching methodology and evaluation system for the course:

Teaching methodology: Instruction: This method recognizes that students have different learning styles, abilities, and backgrounds, and aims to create a learning environment that accommodates these differences.

Evaluation System –

- A. **Internal Assessment (60 Marks)-** Formative Continuous Assessment [Continuous Assessment; Note Book (30 Marks), Viva Voce (20 Marks), Attendance (10 Marks)]
- B. End-Semester Exam (40 Marks)- Summative Assessment.

4. Course Content:

Course Name: BASICELECTRONICS ENGINEERING LAB

Course Code: ESC-EC 191 Hours per Week: 0L: 0T: 2P

Module	Topics	13P
1.	 Familiarization with electronic component and measuring instruments: Familiarization with Electronic components such as Resistors, Capacitors, Diodes, Transistors etc. Familiarization with measuring equipment like Multimeter, Trainer-kit, CRO, Signal Generator etc. 	2P
2.	 Experiment on Diode and Diode Circuits: Study on V-I characteristics of Junction Diode. Realization of positive and negative Clippers circuit using Diode. Realization of Clamper circuit using Diode Realization of Bridge Rectifier using Diode Realization of Voltage regulator using Zener Diode. 	5P
3.	 Experiment on BJT and MOSFET: Study of V-I Characteristics of Bipolar Junction Transistors (BJT). Study on V-I Characteristics of MOS-Field Effect Transistors (MOSFET). 	2P
4.	Experiments on OP-AMP circuit: 1. Realization of Inverting and non-inverting Op-amp amplifier 2. Realization of Op-amp Adder	2P
5.	Experiment on Digital Logic: 1. Truth table verification of basic and universal logic gates 2. Logic verification and design of XOR using NAND-Gate	2P

Textbooks

- Electronic Principles, by Sanjay Sharma, S.K.Kataria & Sons
- Integrated Electronics by Millman & Halkias

Reference Books

- Electronics Devices & Circuits by Salivahanan
- Electronics Circuits by Schilling & Belove

6. Course Outcomes (CO):

After going through this course the Students will be able to:

Course	Details/Statement	Action	Knowledge
Outcomes		Verb	Level
ESC-EC 191.1	Understand different electronic components and measuring equipments	Explain	Understand
ESC-EC 191.2	Select and Examine the characteristics of P-N Junction Diode, BJT and MOSFET	Identify, Select	Examine
ESC-EC 191.3	Apply and execute diode clipper, clamper and rectifier circuit	execute	Apply
ESC-EC 191.4	Apply and implement Zener-diode Circuit	Implement	Apply
ESC-EC 191.5	Implement and examine op-amp circuit	Implement	examine
ESC-EC 191.6	Implement and analyze digital circuit using logic gate	Implement	Analyze

7. Mapping of Course Outcomes (CO) to module / course content

Module	CO1	CO2	CO3	CO4	CO5	CO6
1	3	-	-	-	-	-
2	-	2	3	2	-	-
3	-	3	-	-	-	-
4	-	-	-	-	3	-
5	-	-	-	-	-	3

8. Mapping of the Course Outcomes (CO) to Program Outcomes (PO)

					-	,	U					
	PO1	PO	PO1	PO1	PO1							
		2	3	4	5	6	7	8	9	0	1	2
CO1	2	2	1	2	2	-	-	-	-	-	-	2
CO2	3	3	1	2	2	-	-	-	-	-	-	2
CO3	3	3	1	2	2	-	-	-	-	-	-	2
CO4	3	3	1	2	2	-	-	-	-	-	-	2
CO5	3	3	1	2	2	-	-	-	-	-	-	2
CO6	3	3	1	2	2	-	-	-	-	-	-	2

	PSO1	PSO2	PSO3	PSO4
CO1	1	2	1	1
CO2	2	1	1	1
CO3	1	1	1	1
CO4	1	2	1	1
CO5	1	1	1	1
CO6	1	2	1	1



Course Name: Introduction to Computer Hardware and Software Lab Course Code: ESC-CS 191 (Semester- I)

Course Broad Category: Engineering Science

1. Course Prerequisite:

Basic knowledge of computers and general mathematical operations.

2. Course Learning Objectives:

- i. To introduce students to the fundamental concepts of computer hardware and software.
- ii. To equip students with knowledge and practical skills essential for understanding modern computing systems and networks.
- iii. To provide an overview of emerging computing technologies and their applications in various domains.

3. Teaching methodology and evaluation system for the course:

Teaching methodology –Lectures and Presentations, Interactive Discussions and Case Studies.

Evaluation System -

- A. Internal Assessment (60 Marks)- Formative Continuous Assessment
- D. End-Semester Exam (40 Marks)- Summative Assessment.

4. Course Content:

Course Name: Introduction to Computer Hardware and Software Lab

Course Code: ESC-CS 291 Hours per Week: 0L: 0T: 4P

Unit	Content
Unit 1: Experiments on Dismantling PCs (1P)	 Dismantling the System Unit: Recognize all major components inside a PC, describe the function of each component, and define the relationship of internal components. Networking Hardware: Familiarize with the basic hardware required for networking.
Unit 2: Basics of CLI and GUI (3P)	 - **a) Command Line Interface (CLI)**: - Learn the main set of commands, shortcut keys, switches, path, current directory, and parent directory.

- Understand the power and flexibility of CLIs, especially for network and system administrators. - Perform tasks such as: 1. Making a list of all files in a photo folder and saving it in a text document. 2. Copying all files changed in the last week to a flash drive. 3. Deleting all files ending in ".doc". 4. Renaming every file ending in ".txt" to ".doc". 5. Other similar tasks. - **b) Graphical User Interface (GUI) and Google Drive**: - Perform similar tasks using a GUI and compare them with CLI operations. - Execute tasks using Google Drive. - **a) Basic Document Creation**: Unit 3: Report - Learn all shortcut keys. Formatting using MS Word and - Design and create a Bio-Data or a technical report, including Google Docs (2P) mathematical functions. - **b) Project Report Formatting**: - Given a project report in PDF format, transfer it to word processor software (MS Word or Google Docs). - Format the document to include: - Title page - Specified paragraph and page formatting (page size, orientation, line spacing, font type, font size, indent, bullets, paragraph formatting) - Acknowledgement page - Table of contents page - List of figures page - List of tables page - Bibliography and references - Distinct headers for each chapter - Page numbering in Roman numerals for initial pages and standard numbering from the first chapter - Check for spelling errors and make corrections. Unit 4: Content - **a) Creating Presentations**: Presentation using - Prepare presentations on topics such as "Impact of Social Presentation Media on Youth" and "Emerging Trends in Mobile Software (MS Technology". PowerPoint and - Include slide animations, slide transitions, sound recordings, Google Slides) (2P) slide timings, and customer feedback videos. - Export the presentation as a video or save it as a slideshow. - Prepare handouts for the audience.

	 Debug one or two C programs line by line. Understand and explain the effect of the execution of individual lines at the memory level.
Unit 7: Fundamentals of C Programming (4P) (Optional for Group A)	 - **a) Understanding C Programs**: - Learn the structure of a C program, the necessity of preprocessor directives, and how to compile and execute simple C programs. - **b) Debugging C Programs**:
Unit 6: Data Capture using Google Forms (1P)	 - **a) Creating a Quiz**: - Design and create a quiz using Google Forms. - **b) Data Forms**: - Create data forms for: - Event registration - Event feedback - Customer feedback/satisfaction on a product or service - Order requests
Unit 5: MS Excel and Google Sheets (3P)	following: - Add audio and video effects. - Apply various color schemes. - Apply various animation schemes. - Apply slide show settings. - **c) Using Google Slides**: - Perform similar tasks using Google Slides. - **a) Basic Operations in MS Excel and Google Sheets**: - Create a worksheet containing roll numbers and marks in 2 subjects for 50 students. - Calculate results and grades: - A student is declared as PASS if they score 40 or more in both subjects, otherwise FAIL. - All FAILED students are given Grade IV. - For PASSED students, the grade is assigned as follows: - Average >= 60 → Grade I - <60 but >=50 → Grade II - <50 but >=40 → Grade III - **b) Similar Tasks Using Google Sheets**: - Perform the same tasks using Google Sheets.
	- **b) Wild-Life Presentation**: - Make a presentation on "Wild-Life" and apply the

Text & References Books:

- Introduction to Computer Science; Pearson publication
- Fundamentals Of Computers by V. Rajaraman, PHI Learning
- Fundamentals of Computersby E Balagurusamy, McGraw Hill Education (India)

6. Course Outcomes (CO):

After going through this course the Students will be able to:

Course	Details	Action Verb	Knowledg
Outcomes			e Level
ESC-CS	Identify and describe the major components of a	Identify,	Knowledge,
191.CO1	computer system and understand their	Describe	Comprehensio
	interrelationships		n
ESC-CS	Demonstrate proficiency in using both CLI and GUI	Demonstrate,	Application,
191.CO2	to perform various tasks.	Perform	Analysis
ESC-CS 191.CO3	Design, format, and create professional documents and presentations, including project and technical reports, using MS Word, Google Docs, MS PowerPoint, and Google Slides.	Design, Format, Create, Integrate	Application, Synthesis, Evaluation
ESC-CS	Perform basic operations in MS Excel and Google	Perform,	Application,
191.CO4	Sheets for data analysis and grade calculation.	Calculate	Analysis
ESC-CS	Perform basic operations in MS Excel and Google	Perform,	Application,
191.CO5	Sheets for data analysis and grade calculation.	Calculate	Analysis
ESC-CS	Understand and debug simple C programs,	Understand,	Knowledge,
191.CO6	demonstrating the knowledge of program structure, preprocessor directives, and memory-level execution.	Debug	Application, Comprehensi on

7. Mapping of course outcomes to module / course content

Module	CO1	CO2	CO3	CO4	CO5	CO6
1	3	-	-	-	-	-
2	-	3	-	-	-	-
3	-	-	3	-	-	-
4	-	-	3	-	-	-
5	-	-	-	3	-	-
6	-	-	-	-	3	
7	-	-	-	-	-	3

8. Mapping of the Course outcomes to Program Outcomes (PO)

	PO1	PO2s	PO3	PO4	PO5	PO6	PO7	PO8	PO9	PO10	PO11	PO12
CO1	3	2		1	2			1		2	2	2
CO2	3	3	2		3				2		2	
CO3			3					1		3		3
CO4	1	3		2	3			1	2	3	2	2
CO5		3	3		3						2	2
CO6	1		3	2				1	2	2	3	3

	PSO1	PSO	PSO3	PSO4
		2		
CO1	3	3	3	3
CO2	2	2	2	3
CO3	3	2	3	3
CO4	3	2	2	2
CO5	3	3	3	3
CO6	3	3	3	3



Course Name: ENGINEERING GRAPHICS

Course Code: ESC-ME 191 (Semester- I)

Course Broad Category: Engineering Science

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1. Course Prerequisite: Understanding of basic geometric concepts, such as points, lines, angles, and planes.

2. Course Learning Objectives:

- Introduction to engineering design and its place in society
- Exposure to the visual aspects of engineering design
- Exposure to engineering graphics standards
- Exposure to solid modeling

3. Teaching methodology and evaluation system for the course:

Teaching methodology – Lectures and Presentations, Interactive Discussions and Case Studies, Guest Lectures and Field Visits.

Theoretical Instruction, Hands-on Practice, Assessment & Feedback, Interactive Learning Methods, Documentation & Reporting.

Evaluation System -

Section 1: Practical Continuous Internal Assessment (PCIA) - 60 Marks

Includes practical performance, reports, and viva voce after each experiment. Throughout the Semester

Section 2: Practical End Semester Examination (PESE) - 40 Marks

Final comprehensive practical examination covering the entire syllabus, at the end of the semester

4. Course Content:

Course Name: Engineering Graphics
Course Code: ESC-ME191/ ESC-ME 291

Hours per Week: 0L: 0T: 4P

Module	Topics	Practical (P)
	INTRODUCTIONTOENGINEERINGDRAWING	
1	Principles of Engineering Graphics and their significance, usage of Drawing in struments, lettering, Different types of line sand theiruse, Dimensioning.	4
2	SCALES	4
2	Plain scale, Diagonal scale and Vernier Scales.	4
	GEOMETRICALCONSTRUCTIONANDCURVES	
3	Construction of polygons, Conic sections including the Rectangular Hyperbola(General method only); Cycloid, Epicycloid, Hypocycloid, Involute, Archemedian Spiral.	4
	PROJECTIONOFPOINTS,LINES,SURFACES	
4	Principles of Orthographic Projections-Conventions - 1st and 3rd angle projection, Projections of Points and lines inclined to both planes; Projections of planes (Rectangle, pentagon, Hexagon etc.)inclined Planes	8
	-Auxiliary Planes.	
	PROJECTIONOFREGULARSOLIDS	
<i>-</i>	Projectionofsimpleregularsolids,viz.prisms,cubes,cylinders,pyr	4
5	amids, cones, tetrahedrons, spheres, hemi-spheres etc. SECTIONS AND SECTIONAL VIEWS OF RIGHT	4
6	ANGULAR SOLIDS Sectionofsolids; section by perpendicular planes; sectional views; true shapes of sections.	4
	DEVELOPMENTOFSURFACES	
7	Development of surfaces of Right Regular Solids - Prism, Pyramid, Cylinder and Cone.	4
8	ISOMETRIC PROJECTIONS Principles of Isometric projection – Isometric Scale, Isometric Views, Conventions; Isometric Views of lines, Planes, Simple and compound Solids; Conversion of Isometric Views to Orthographic Views and Vice-versa, Conventions.	4

	OVERVIEWOFCOMPUTER GRAPHICS	
9	Listing the computer technologies that impact on graphical	4
	communication, Demonstrating knowledge of the theory of	
	CAD software [such as: The Menu System, Toolbars	
	(Standard, Object Properties, Draw, Modify and Dimension),	
	Drawing Area (Background, Crosshairs, Coordinate System),	
	Dialog boxes and windows, Shortcut menus (Button Bars), The	
	Command Line (where applicable), The Status Bar, Different	
	methods of zoom as used in CAD, Select and erase objects.;	
	Isometric Views of lines, Planes, Simple and compound	
	Solids];	

- 1. Pradeep Jain, Ankita Maheswari, A.P. Gautam, Engineering Graphics & Design, Khanna Publishing House
- 2. Bhatt N.D., Panchal V.M. & Ingle P.R., (2014), Engineering Drawing, Charotar Publishing House
- 3. Agrawal B. & Agrawal C. M. (2012), Engineering Graphics, TMH Publication
- 4. Shah, M.B. & Rana B.C. (2008), Engineering Drawing and Computer Graphics, Pearson Education
- 5. Narayana, K.L. & P Kannaiah (2008), Text book on Engineering Drawing, Scitech Publishers
- 6. Corresponding set of CAD Software Theory and User Manuals
- 7. Pal & Bhattacharya (2006), Engineering Drawing, 6/e, Viva Books Private Limited,

6. Course Outcomes (CO):

At the end of the course, the students will be able to-

COs	Description	Action Verb	Knowled
			ge level
ESC- ME	UNDERSTAND the basic concept of lines and dimension	UNDERSTAND	Understan
191.1	for engineering drawing.		d
	INTERPRET and construct scales (plain, vernier &	INTERPRET	Understan
ESC- ME	diagonal) and curves (ellipse, parabola, hyperbola,		d
191.2	cycloids, involute, spiral).		
ESC-	CONSTRUCT orthographic projections of points, lines,	CONSTRUCT	Apply
ME 191.3	planes, solids.		
ESC-	ILLUSTRATE surface development and section of solids.	ILLUSTRATE	Apply
ME 191.4	indeed it is a section of solids.		

ESC- ME 191.5	APPLY the concept of isometric projection.	APPLY	Apply
ESC- ME	UNDERSTAND the basic concept of Auto CAD in engineering drawing.	UNDERSTAND	Understan d

7. Mapping of the Course Outcomes (CO) to Program Outcomes (PO)

COs	РО	РО	РО	РО	PO	РО						
	1	2	3	4	5	6	7	8	9	10	11	12
ESC-ME 291.1	3	1	1	-	-	-	-	-	-	-	-	1
ESC-ME 291.2	3	1	1	-	1	1	-	1	1	1	-	1
ESC-ME 291.3	3	1	1	-	1	1	-	1	1	1	-	1
ESC-ME 291.4	3	1	1	-	-	-	-	-	-	-	-	1
ESC-ME 291.5	3	1	1	-	1	1	-	1	1	1	-	1
ESC-ME 291.6	3	1	-	-	1	1	-	-	-	-	-	1
AVG.	3	1	1	-	1	1	-	1	1	1	-	1

8. Mapping to Program Specific Outcome (PSO)

	PSO1	PSO	PSO3	PSO4
		2		
CO1	2	1	2	1
CO2	1	1	2	1
CO3	1	2	1	1
CO4	1	1	2	1
CO5	1	1	1	2
CO6	1	1	2	1

Following is the list of drawing instruments that required for Engineering Drawing:

1. Drawing Board

- 2. Mini drafter/Set-squares(45° – 45° & 60° – 90°),T-square
- 3. Protractor(180°,360°)
- 4. Scales(Plain, Diagonal)
- 5. Compass(Small and Large)
- 6. Divider(Small and Large)

*** End of Syllabus***